# Energy

### 1. Energy in food

Energy stored in food can be released by combustion (burning) or by respiration in our cells. The labels on packets of food show how much energy is available from the food. In some instances food energy is measured in kilojoules - mostly by the scientific community - though some food packaging also gives kilojoule (kJ) values. 1 kilocalorie = 4.2 kilojoules.

### 2. Energy stores and transfers

Energy Store	Definition		
Kinetic energy	All moving things have kinetic energy. The amount of kinetic energy an object has depends upon: • the mass of the object • the speed of the object.		
Thermal energy	<b>Thermal energy</b> is energy possessed by an object or system due to the movement of particles. The faster the movement of the particles the greater the thermal energy and temperature.		
Elastic potential energy	Some objects can change shape reversibly, e.g. rubber balls, springs and elastic bands. When a rubber ball is stretched or squashed, it can regain its shape again. Elastic potential energy is stored in stretched or squashed materials.		
Gravitational potential energy	<ul> <li>When an object is moved higher, it gains gravitational potential energy. The amount depends upon:</li> <li>the mass of the object</li> <li>the extra height it gains</li> <li>the gravitational field strength (on Earth this is 9.81 m/s<sup>2</sup>. it is often rounded to 10 m/s<sup>2</sup>).</li> </ul>		
Electrostatic	Some objects carry electrical charges and create electric fields. These charged objects can exert forces on each other. You get an electric current when charged particles move through a wire.		
Magnetic	Some objects can be magnetised and create magnetic fields. They can exert forces on other magnetised objects, or on magnetic materials.		
Chemical	The energy contained in a chemical substance, such as food, petrol or a battery.		
Atomic	The energy contained in a nuclear substance, such as stars, nuclear weapons and fuel for nuclear power stations.		

#### 3. Heat and temperature

A thermometer is used to measure the temperature of an object

The temperature of an object is a measure of the average kinetic energy per particle of the object. It is measured in degrees Celsius. Note that the unit of temperature is written as °C. (not °c or oC).

All objects contain internal energy. Some of this is due to the movement of the particles in the object. When an object is heated, its particles move more vigorously and its internal energy increases. Unless the object changes state (e.g. melts or boils), its temperature will increase.

4. Conservation of energy. Energy can be stored or transferred, but it cannot be created or destroyed. This means that the total energy of a system stays the same. Sankey diagrams show the conservation of energy and how energy stores empty and fill.



Sankey diagram for a petrol engine

The total amount of energy never changes, it just changes the type of store it is in.

Energy Transfer	Definition	
Heating (by particles)	Energy is transferred from a hotter object to a cooler one. This can be done by conduction or convection.	
Working Mechanically	Energy can be transferred mechanically through the movement of the parts in machines, and when the motion or position of an object changes. Sound waves and seismic waves (formed during earthquakes) are mechanical waves that transfer energy through materials and from place to place.	
Working Electrically	Energy is transferred when an electrical circuit is complete.	
Heating (by radiation)	Visible light, infrared light, microwaves and radio waves are forms of radiation. They are carried by waves (although unlike sound, these are not mechanical waves and can travel through empty space).	

#### 7. Power

Power is the rate at which energy is transferred.

The unit of power is the watt, which has the symbol W. 1 W is 1 J per second.

$$power(W) = \frac{energy(J)}{time(s)}$$

$$er(W) = rac{energy(J)}{time(s)}$$



 $energy (kWh) = power (kW) \times time (h)$ 

The cost of the energy used can be calculated:  $cost = energy (in kWh) \times cost of 1 kWh$ 

### 8. Work Done

By doing "work" you are transferring energy

work done  $(I) = force(N) \times distance(m)$ 

When a force causes a body to move, work is being done on the object by the force. Work is the measure of energy transfer when a force (F) moves an object through a distance (d). So when work is done, energy has been transferred from one energy store to another, and so:

energy transferred = work done

Energy transferred and work done are both measured in joules (J).

## 4 & 5 Thermal equilibrium

If there is a difference in temperature between two objects, energy is transferred from the hotter object to the cooler one. This will continue until both objects are at the same temperature. When they are at the same temperature, we say that they are in thermal equilibrium, and there is no overall transfer of energy any more between the two objects.

Energy can be transferred from a hot object to a cooler one by:

Description

end.

When a substance is

more vigorously. The

heated, its particles gain

internal energy and move

particles bump into nearby

particles and make them

vibrate more. This passes

internal energy through the

from the hot end to the cold

The particles in liquids and

gases can move from place

happens when particles with

liquid or gas move, and take

a lot of thermal energy in a

the place of particles with

transferred from hot places

to cold places by convection.

All objects transfer energy to

infrared radiation. The hotter

infrared radiation it gives off.

No particles are involved in

radiation, unlike conduction

to place. Convection

less thermal energy.

their surroundings by

an object is. the more

and convection.

Thermal energy is

substance by conduction,

- conduction
- convection
- radiation

Type of

transfer

(if the

particles

Conduction

are in fixed

Convection

(if the

move)

particles

are free to

Radiation

positions)

heat



electrical energy. Some pocket

calculators use solar cells. and

of solar cells on house roofs.

you may have seen large panels

costs and no

gases are

produced.

harmful polluting

so the cost of their electricity is high.

Solar cells are not reliable as they do not

work at night and not as well when it is

cloudy.

Energy is transferred from warm homes to the outside by: conduction through the walls, floor, roof and windows, and radiation from the walls, roof and windows. Insulation such as double glazing, carpets and reflective foil reduce the energy transferred.

25% 25%	Solar Cells	No
25% 15%	Collo	~ 4

Yes