Toynbee Curriculum KS3 Topic Summaries

SCIENCE

Toynbee School



Scheme of Learning: Lab Skills

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Lab Skills	Particles and Separation Techniques	Forces	Cells and Organisation	Elements and the Periodic Table	Energy	Health and Human Body	Chemical Reactions	Electricity and Magnetism	Reproduction
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Topic Overview:

This is the introductory topic all year 7 pupils will complete to enable them to carry out future practical activities in science safely.

Lesson Sequence:

We teach the basic lab safety rules and hazard symbols on chemicals and how to correctly use a Bunsen burner and other key pieces of lab equipment.

We then give an overview of the scientific method with a focus on accuracy and precision and how to collect and display data.

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Se	quence of Lessons:	Res	Sources:
1	Lab Safety	1	A variety of chemicals with hazard symbols on.
2	Bunsen burners	2	n/a
3	Equipment	3	Salt.
4	Accuracy & Precision	2/	Blue liquid and red liquid. Measured out precisely
5	Observing		(doesn't matter the volume so long as you tell
6	Displaying Data 1	4	the teacher) e.g. 17ml red, 57ml blue. (so they will need a different sized measuring cylinder for
7	Displaying Data 2		each colour.
8	Summary lesson	5	Magnesium, hydrochloric acid, vinegar, calcium
0	A CIENTI COM	J	carbonate, copper sulphate
	A HOS	6	n/a
		1	n/a
		8	n/a
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Supportive Reading:			L.	m	0-	o (
Comprehension activity	твс						
6	~	\square			0		2
Assessment:			~	~			
Knowledge:	Multiple choi	ce questions.					
Application of Knowledge:	n/a for introd	luctory topic					
	10 M						-

Scheme of Learning: Particles & Separation Techniques

Topic Sequence:

-101	2	3	134 73	5	6	1	8	9	o 10
Lab Skills	Particles and Separation Techniques	Forces	Cells and Organisation	Elements and the Periodic Table	Energy	Health and Human Body	Chemical Reactions	Electricity and Magnetism	Reproduction

Topic Overview:

The national curriculum requires that we teach the following:

- the properties of the different states of matter (solid, liquid and gas) in terms of the particle model.
- changes of state in terms of the particle model
- mixtures, including dissolving
- diffusion in terms of the particle model
- simple techniques for separating mixtures: filtration, evaporation, distillation and chromatography

Lesson Sequence:

We start with an introduction to particles and states of matter. We build on that to look at changes of state, then the movement of particles and finally separation techniques.

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Se	quence of Lessons:	Res	Sources:
1	States of Matter	1	Selection of solids, liquids and gases (include sand,
2	The Particle Model		hair gel/jelly)
3	Changes of State	2	Lego blocks
4	Boiling Vs Evaporation	3	Stearic acid warmed in water bath
5	Melting & Freezing		Sulphuric acid, Copper oxide, Post-its for labelling
6	Diffusion		dishes
1	Solubility	5	Petri dishes, cubes of lard, butter, chocolate
8	Saturation	7	Demo: Potassium permanganate crystals
9	Filtering	6	Class set: Equal sized cubes of agar jelly containing
10	Distillation	0	phenolphthalein, various concentrations of HCl
11	Chromatography	1	Liquids: Ethanol, copper sulphate solution
12	Revision	4	Solids: Coffee, sugar, salt
13	Assessment	8	Sugar
ÿ.	a cleipil (9	Rock salt mixture
	A HOS CO	10	Inky water, delivery tubes & conical flasks
		11	Chromatography paper strips, pens for testing.
		12	n/a
	425	13	Worksheets in shared folder.
		-	FUUUL 0

Supportive Reading:				m	0-(_)-0	> ((
Comprehension activity	твс						
6	A						2
Assessment:	///	129					
Knowledge:	Multiple choic	ce questions.					
Application of Knowledge:	Describe the s	structure and prop	erties of the	three state	es of matter.		

Scheme of Learning: Forces

Topic Sequence: 3 1 2 4 5 6 7 8 9 10 **Elements and** Electricity **Particles and Cells and** Chemical **Health and** Lab Skills Separation Forces the Periodic Energy and Reproduction Organisation Human Body Reactions **Techniques** Table Magnetism

Topic Overview:

The Forces topic aims for pupils to develop their understanding of why objects behave in the world around them; from why they stay on the surface of the Earth, to why their car will speed up and slow down. This topic aims to link the importance of understanding forces with its impact on sport. We use the context of Olympic track cycling to gain an understanding of how forces underpin the design of a bike, as well as how changing body position and clothing can affect speed and increase the chance of winning an Olympic medal.

This topic builds on the KS2 idea that forces are pushes and pulls and begins to identify some basic forces. We build the foundations which will later be developed in the 'motion and pressure' topic in year 8 and 'Forces and interactions' and 'Forces and motion' in KS4 (GCSE).

Lesson Sequence:

Pupils will start by being able to list the different types of forces, and the direction that they act. They will need to understand if the forces are contact or non contact and how they interact with an object. Students will then develop this knowledge by describing if forces in opposite directions are balanced or unbalanced and to calculate a resultant force. Once they can identify resultant forces they will then be able to apply this knowledge to explain how resultant forces affect the object. We start by looking at how it affects the motion of an object (speed and direction). We then focus on two specific resistive forces, drag and friction, and the ways we can reduce them. We then move on to looking how resultant forces cause a change of shape. Finally the pupils look at the relationship between weight and mass. Pupils will be assessed by being asked to explain why changing the body position, tyre thickness or shape of a helmet can change a cyclist's speed.

Se	equence of Lessons:	Reso	Durces:
1 2	Intro to forces (contact and non-contact) Representing forces	1	Newton meters (Range high and low resolution) Objects of different weights. Work sheet 1: Forces definition match
3	Resultant forces Drag	2	Work Sheet 2: <u>Table of results</u> Large Squared (graph) Paper. Worksheet 1: Free body diagram practice.
4		3	Worksheet 1: Resultant force practice 1 Worksheet 2: Resultant force practice 2
5	Friction		Equipment.
6	Changing shape		Hair dryer x 4
7	Weight and Gravity		Small dynamics trolly x 8
8	Forces assessment		Plasticine. 100g x 8
		5	Equipment: Friction Block and Newton Meter X 8 Table of results
		6	Equipment:1 x spring 1x strawberry lace, 1 x clamp stand, a meter ruler, 2 x clamps and 2 x bosses. Table of results and graph to annotate.
		1	Graph paper or print slide 10 to allow them to plot the graph.
		8	Print off assessment sheet.
Sı	pportive Reading:		$\gamma = \gamma =$
Ca	mprehension activity TBC		AS AN ELSA

Assessment:	
Knowledge:	20 question multiple choice quiz
Application of Knowledge:	Explain why changing the body position, tyre thickness or shape of the helmet can change a cyclist's speed.

Scheme of Learning: Cells & Organisation

Topic Sequence:

<u>1</u>	2	3	4	5	6	>> 1	8	9	10
Lab Skills	Particles and Separation Techniques	Forces	Cells and Organisation	Elements and the Periodic Table	Energy	Health and Human Body	Chemical Reactions	Electricity and Magnetism	Reproduction

Topic Overview:

The national curriculum requirements for this topic are to cover the following:

- cells as the fundamental unit of living organisms, including how to observe, interpret and record cell structure using a light microscope
- the functions of the cell wall, cell membrane, cytoplasm, nucleus, vacuole, mitochondria and chloroplasts
- the similarities and differences between plant and animal cells
- the role of diffusion in the movement of materials in and between cells
 the structural adaptations of some unicellular organisms
- the hierarchical organisation of multicellular organisms: from cells to tissues to organs to systems to organisms
- the structure and functions of the human skeleton, to include support, protection, movement and making blood cells
- biomechanics the interaction between skeleton and muscles, including the measurement of force exerted by different muscles
- the function of muscles and examples of antagonistic muscles.

Lesson Sequence:

Starting with learning the new skill of using a microscope, we move onto learning the parts and functions of animal and plant cells and how they differ from microbes and specialised cells. We then build up from cells to tissues to organs then organ systems. We focus on the muscle and skeletal systems as an example of an organ system.

Seq	juence of Lessons:	Res	ources:
1	Using a microscope	1	Microscopes, Pre-prepared microscope slides
2	Animal cells		Microscopes, Slides, Cover slips, Cotton buds, Disinfectant pot for
3	Plant cells	Z	used slides Methylene blue dye
4	Specialised cells	3	Microscopes, Slides, Cover slips, Iodine
5	Microbes	-	Onions, Knives, White tiles
6	Cells to Organisms	4	Specialised cells info sheets and table for pupils to complete (laminated in cabinet)
7	Skeletal system	5	Microbe fact sheets and question sheet
8	Muscles & chicken dissection	•	(laminated in cabinet)
9	Antagonistic muscles	6	n/a
10	Assessment	7	Skeleton model, Joints fact sheet, (laminated in cabinet),
		8	Chicken wings, Dissection tools, Bin bag for remains Disinfectant and extra paper towels to clean up after.
		9	Arm muscle template printed onto card. Split pins, String, Enough scissors if not enough in classroom
		10	In folder

Supportive Reading:	
Comprehension activity	Unicellular organism comprehension sheets
Assessment:	
Knowledge:	20 question multiple choice knowledge test
Application of Knowledge:	Extended writing answer comparing plant and animal cells.

Scheme of Learning: Elements and The Periodic Table

Topic Sequence:

901	2	3	4	5	6	1	8	9	10
Lab Skills	Particles and Separation Techniques	Forces	Cells and Organisation	Elements and the Periodic Table	Energy	Health and Human Body	Chemical Reactions	Electricity and Magnetism	Reproduction

Topic Overview:

The national curriculum requires that we teach the following:

- Differences between atoms, elements and compounds
- Chemical symbols and formulae for elements and compounds.
- The concept of a pure substance.
- The varying physical and chemical properties of different elements.
- The principles underpinning the Mendeleev Periodic Table.
- The Periodic Table: periods and groups; metals and non-metals.
- How patterns in reactions can be predicted with reference to the Periodic Table.
- The properties of metals and non-metals
- The chemical properties of metal and non-metal oxides with respect to acidity.

Lesson Sequence:

We start with an introduction to the Periodic Table, including uses, properties and trends. We then use this knowledge to identify differences in elements, compounds and mixtures. Once pupils have understood this we then explore the uses and patterns found in groups 1,7 and 0 in more depth.

Sequ	ence of Lessons:	Res	Resources:				
1	The Periodic Table	1	Print table for pupils to complete				
2	Uses of Elements	9	Info slides for elements in prep room drawers, squares of				
3	Trends in The Periodic Table	2	plain paper				
4	Development of The Periodic Table	T	Print table for pupils to complete. Selection of metals (Mg, Al,				
5	Compounds and Mixtures	3	Zn, cobalt, nickel etc), 1 basic electrical circuit set up (bulb,				
6	Pure an Impure Substances		ammeter, wires, battery), tray of water, set of bar magnets				
h	Group 1 Elements	4	Print timeline slide and pictures to cut and stick slide. Both on				
8	Group 7 Elements		A3.				
9	Group 0 Elements	5	Print worksheet for pupils to complete. Possible demo sandy water, sieve, filter paper, funnel.				
10	Revision	7	Print graph worksheet. Salt water and pure water (for use				
11	Assessment	- 6	with Bunsen burners and thermometers).				
	A HURS	1	Demo: Li, Na, K in water trough, UI.				
		8	None				
		9	None				
	123	10	Revision				
	A MANIC) 11	Assessment sheets in shared folder.				
Supp	ortive Reading:						
Comp	rehension activity TBC						

	-					
0	X N	1 73		0		10
Assessment:		LES	0.0		0	
Knowledge:	Multiple choi	ice questions.				
Application of Knowledge:	Compare the	trends and uses of	group 7 and grou	o 1 elements.		
	0					0 0

Scheme of Learning: Energy

Topic Sequence:

1	2	3	4	5	6	1	8	9	10
Lab Skills	Particles and Separation Techniques	Forces	Cells and Organisation	Elements and the Periodic Table	Energy	Health and Human Body	Chemical Reactions	Electricity and Magnetism	Reproduction

Topic Overview:

Energy is a new topic at KS3 as no content has been covered at KS2. This topic looks at energy as a mathematical concept, the idea that energy allows us to work out if something could happen.

The topic includes:

Calculation of fuel uses and costs in the domestic context

Energy changes and transfers - simple machines; heating and thermal equilibrium; changing motion, dropping an object, stretching a spring, metabolism of food, burning fuels.

Changes in systems - energy as a quantity that can be quantified and calculated; the total energy has the same value before and after a change

Lesson Sequence:

We begin with a summary of the different stores and pathways. Then move into the detail of how the stores can be calculated and the processes involved in the transfer of energy (pathways).

We cover the calculations of work done, power and the cost of electricity. We also look at the means of representing energy transfers in Sankey diagrams.

Finally, we look at the resources used to generate the electricity we use in our homes, including the advantages and disadvantages for each resource.

Please note: some of the lessons will take more than the 1 hour lesson slot. Please account for this in your advanced planning.

Sec	juence of Lessons:	Res	ources:
1	Energy stores	1	Energy stores table
2	Chemical stores	2	Combustion of food experiment – different foods, tin lids, mounting pins + basic lab equipment
3	Pathways	3	Work done calculations sheet
4	Conduction and convection	4	Conduction experiment – metal rods, drawing pins, Vaseline + basic lab equipment Convection experiment – potassium permanganate 'tea bags' + basic lab equipment Chimney demo
5	Heat and temperature	5	Image sheet for defining vocabulary
6	Sankey diagrams	6	Sankey diagrams sheet
7	Non- renewable resources	1	Power station worksheet
8	Renewable resources	8	Renewables table sheet Resources posters (laminated)
9	Power and cost	9	Power calculations
10	Assessment	10	Quiz sheet Assessment sheet

Supportive Reading:	
Comprehension activity	ТВС
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Assessment:	
Knowledge:	20 question multiple choice quiz
Application of Knowledge:	Extended writing task evaluating the use of an energy resource

Scheme of Learning: Health & the Human Body

Topic Sequence:

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< <u>°</u> 01 ∕	2	3	4	5	6	1	8	9	10
SKILLS	Particles and separation	Forces	Cells and organisation	Atomic structure and the Periodic Table	Energy	Health and human body	Chemical reactions	Electricity and magnetism	Reproduction

Topic Overview:

The national curriculum requirements for this topic are to cover the following:

- Content of a healthy human diet: carbohydrates, lipids (fats and oils), proteins, vitamins, minerals, dietary fibre and water, and why each is needed
- · Calculations of energy requirements in a healthy daily diet
- · The consequences of imbalances in the diet, including obesity, starvation and deficiency diseases
- The tissues and organs of the human digestive system, including adaptations to function and how the digestive system digests food (enzymes simply as biological catalysts)
- the structure and functions of the gas exchange system in humans, including adaptations to function
- The mechanism of breathing to move air in and out of the lungs, using a pressure model to explain the movement of gases, including simple measurements of lung volume
- The impact of exercise, asthma and smoking on the human gas exchange system
- The effects of recreational drugs (including substance misuse) on behaviour, health and life processes.

Lesson Sequence:

Starting with the respiratory system, we learn the parts and functions of different organs. We then move on to a healthy & unhealthy diet before learning about the digestive system. We focus on the structure and function of these two system.

Seq	uence of Lessons:	Res	OURCES:
1	Hierarchy recap & Function of breathing/respiration	<u>_</u>	Torso including respiratory & digestive systems. CO2 test: limewater, straw, chilled mirrors (ice), stop- clocks. Handouts.
2	Respiratory system structure & function	2 <	Handouts. Torso model (lungs), pluck demo. Rubber tubing, disinfectant, dissecting equipment, gloves for class.
3	Ventilation / Breathing	3	Bell jar (diaphragm), Demo: 2 litre drinks bottles, large basins, clean tubing & disinfectant, cotton wool,
4	Alveoli & diffusion	4	Pink phenolphthalein; 0.5M Hydrochloric acid; Beakers / boiling tubes; Timers
5	Lung health	5	Stop-clock, measuring tape, smokers lung vs non-smokers lung laminated sheets. Smoking machine video clin
6	Healthy diet	6	Laminated nutrient cards; Colouring pencils; true/false sheets for highlighting.
7	Unhealthy diet. Dietary disease. Food tests 1: Fats	• 1	Laminated sheets diet info. Greaseproof paper, pre-weighed crisps or balances (depending on class): low fat, high fat, crackers, rice, etc. 1x1cm squared paper. Instruction sheets. Ethanol & water, pestle &
8	Food tests 2: Starch, glucose, protein.	<u> </u>	mortar, test tubes, filter paper, funnels.
9	Energy in food (linked to L7)	6 8	Test stations with food samples. Starch, spotting tiles, iodine, potato, rice, crisps, cake, etc, Benedicts, kettles/water-bath at 90oC, beakers, boiling tubes, biuret solution, bungs.
10	Digestive system structure & function	g	Corks with pins, balances, puffy crisps, etc, timers, boiling tubes, thermometers, measuring cylinders,
11	Enzyme function	10	Human torso with digestive system. Diagrams to label/sequence. Tape measure.
12	What is a drug: Effects & dangers	6	Demo mesh/orange netting & molymods. Amylase, starch, iodine, visking tubing, benedicts reagent,
13	Effects of alcohol	11	glucose solution, water-bath at 90oC or kettle, Hydrogen peroxide 30% vol, Mn IV oxide (catalyst) potato cubes. Optional tights & fake faeces demo.
14	Assessment	12	Card-sort or whiteboards. Photos of legal/illegal drugs. 4 white powders (see lesson) labelled: A = Table salt; B = citric acid crystals; C = Sugar; D = Talcum powder. UI indicator solution. Benedicts solution.
		13	Reaction times drop-sticks. Data sheets.
		14	Assessment sheets

Supportive Reading:	
Comprehension activity	ТВС
Assessment:	
Knowledge:	20 question multiple choice knowledge test
Annlication of Knowledge	Extended writing answer describing the journey of oxygen & carbon dioxide through the

recniratory system

Scheme of Learning: Chemical Reactions

Topic Sequence:

101	2	3	47	5	6	1	8	9	10
Lab Skills	Particles and Separation Techniques	Forces	Cells and Organisation	Elements and the Periodic Table	Energy	Health and Human Body	Chemical Reactions	Electricity and Magnetism	Reproduction
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Topic Overview:

The national curriculum requires that we teach the following:

- Chemical reactions as the rearrangement of atoms
- Representing chemical reactions using formulae and using equations
- Combustion, thermal decomposition,
- Oxidation and endo and exothermic reactions

Lesson Sequence:

We start with an introduction to physical and chemical changes. We build on this knowledge to look at word and chemical equations, which is then linked with the conservation of mass. Finally we learn a range of chemical reactions (endo and exothermic, combustion, oxidation, thermal decomposition).

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Seq	uence of Lessons:	Res	ources:
1	Physical and Chemical change		Zinc, Copper sulphate, Magnesium, Iron sulphate, Iron
2	Word equations	1	filings, Hydrochloric acid, Sodium bicarbonate,
3	Conservation of mass		Vinegar, Sodium hydroxide
4	Exo and Endothermic reactions	2	Iron, Copper sulphate, Zinc, Silver nitrate,
5	Combustion	-4	Demo : Sulphur, Oxygen, Lithium with water
6	Thermal decomposition	3	Magnesium, Pan-balance
1	Revision	99	Magnesium, Hydrochloric acid, Potassium nitrate,
8	Assessment	4	Sodium carbonate, Sodium hydroxide, Calcium
		5	chloride, Ammonium nitrate
2		5	Candles, plasticine and different size beakers and Demo : combustion apparatus
	B B C	6	Demo: Copper carbonate, Zinc carbonate, Calcium carbonate, Limewater

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Supportive Reading:	1 10		D	TH.		F H Y	
Comprehension activity	твс						
	2	2	0	0	P	2002	3
Assessment:		1.78			9		6
Knowledge:	Multiple choice c	juestions.					
Application of Knowledge:	Compare the con	nbustion reacti	on with tl	ne thermal	decomposition react	ion	
	There are bady			and the second se			and the second s

Scheme of Learning: Electricity & Magnetism

Topic Sequence:

1	2	/4L3-1V	4	5	6	1	8	9	10
Lab Skills	Particles and Separation Techniques	Forces	Cells and Organisation	Elements and the Periodic Table	Energy	Health and Human Body	Chemical Reactions	Electricity and Magnetism	Reproduction

Topic Overview:

Electricity and magnetism has been covered in three of the four years at KS2. Students should have an understanding of simple circuits, including some circuits symbols and understand that magnets can exert a force at a distance on certain materials. In this topic we cover:

Current electricity – students should gain knowledge of current, potential difference and resistance in simple series and parallel circuits.

Static electricity – the is caused by the transfer of electrons between two insulating materials and creates electric fields which can exert a force at a distance.

Magnetism – including magnetic field patterns, the Earth as a magnet, electromagnets and simple d.c. motors.

Lesson Sequence:

We begin with a recap of the information taught at KS2, looking at basic circuits and symbols. This then develops into a more detailed understanding of the concepts of current, potential difference and resistance in electric circuits.

We then look at static electricity caused by the transfer of electrons and demonstrate the effect of electrostatic forces.

Next we look at permanent magnets and identify the magnetic field pattern around a bar magnet. This is then compared to the magnetic field of the Earth.

Finally, we look at how the ideas of the topic can be linked together in electromagnetism and how electromagnets can be useful. *Please note: some of the lessons will take more than the 1 hour lesson slot. Please account for this in your advanced planning.*

Sec	juence of Lessons:	Reso	Resources:					
1	Current	1	Yellow kit bulbs, cells and ammeters, leads					
2	Potential difference	2	Yellow kit bulbs, cells and voltmeters, leads					
3	Series and parallel	3	Worksheet, yellow kit bulbs, cells, ammeters and voltmeters, leads					
4	Resistance	4	Worksheet					
5	Static electricity	5	Van der Graaff worksheet, Demo – balloon, plastic rod and cloth, hole punchings/ground black pepper, Van der Graaff demo					
6	Magnetic materials	6	Ppt slide 7 as a sheet, class set – bar magnets, selection of materials (iron, paper, rubber, nickel, wood, glass, copper, zinc, aluminium), paper clips, thread					
1	Magnetic fields	1	Class set – bar magnets, plotting compasses, iron filings. Demo – navigational compass					
8	Electromagnets	8	Class set – electromagnet wires, iron cores, assorted other cores (e.g. wood, plastic), box of paperclips, variable power supplies, leads, crocodile clips					
9	Electromagnet uses	9	Electric bell diagram Motor kits, variable power supplies (small grey ones are good for this)					
10	Assessment	10	Quiz sheet Assessment sheet					

Supportive Reading:		
Comprehension activity	ТВС	

Assessment:						
Knowledge:	20 question multiple choice quiz					
Application of Knowledge:	Extended writing task evaluating the use of an energy resource					

Scheme of Learning: Reproduction

Topic Sequence:

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<u></u> 1 /	2	3	4	5	6	>- I	8	9	10
Lab Skills	Particles and Separation Techniques	Forces	Cells and Organisation	Elements and the Periodic Table	Energy	Health and Human Body	Chemical Reactions	Electricity and Magnetism	Reproduction
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Topic Overview:

The national curriculum requirements for this topic are to cover the following:

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- reproduction in humans (as an example of a mammal), including the structure and function of the male and female reproductive systems, menstrual cycle (without details of hormones), gametes, fertilisation, gestation and birth, to include the effect of maternal lifestyle on the foetus through the placenta
- reproduction in plants, including flower structure, wind and insect pollination, fertilisation, seed and fruit formation and dispersal, including quantitative investigation of some dispersal mechanisms.

Lesson Sequence:

Starting with the human system, we look at the functions and names of the reproductive organs in males and females. In a logical sequence we then learn about fertilisation, pregnancy, birth and finally puberty. We then move onto reproduction in flowering plants, again learning the functions and names of the relevant organs and then pollination and seed dispersal.

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Sei	juence of Lessons:	Resources:
1	Reproduction systems	Question box and post-it notes. (keep this for
2	Fertilisation in humans	duration of topic)
3	Pregnancy & birth	2 Resources in shared folder
4	Puberty	3 Resources in shared folder
5	The menstrual cycle	4 Resources in shared folder
6	Plant reproduction	5 Resources in shared folder
7	Pollination	6 Cress seed, Petri Dishes, Cotton Wool, Marker Pens,
8	Seed dispersal	
9	Assessment	Resources in shared folder
Y.,		8 Maple seed templates, Paper clips
12	₭ 。 。 。 、 、 、 、	9 In shared folder
1.1.1		

Puberty problem pages L4
20 question multiple choice knowledge test
Extended writing answer describing fertilisation in humans.

Scheme of Learning: Acids and Alkalis

lopic Sequence:									
1	2	3	7 -4	5	6	r(// ^	8	6 9	
Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence	
			11						

Topic Overview:

In year 5, students should study that some materials will dissolve in liquid to form a solution; and explain that some changes result in the formation of new materials, and that this kind of change is not usually reversible, including changes associated with the action of acid on bicarbonate of soda.

This topic then includes:

defining acids and alkalis in terms of neutralisation reactions

the pH scale for measuring acidity/alkalinity; and indicators

reactions of acids with alkalis to produce a salt plus water

Lesson Sequence:

This topic begins by recapping the necessity of completing risk assessments when working with chemicals (covered in the particles topic in year 7). We then move onto looking at the definition of concentration and how this will affect the speed of a reaction (qualitatively). Students will then use a range of indicators to investigate strength and pH. Finally, looking at neutralisation reactions, naming slats and balancing symbol equations.

Please note: some of the lessons will take more than the 1 hour lesson slot. Please account for this in your advanced planning.

S	equence of Lessons:	Reso	Durces:				
1	Hazards and risks	1	A selection of household acids and alkalis in original containers Keyword wordsearch				
2	Concentration	2 A	0.1M HCl with irritant label, 0.1M NaOH with irritant label, 2M HCl with corrosive label, 2M NaOH with corrosive label Flasks with 3 obviously different concentrations of squash Magnesium ribbon				
3	Indicators	3	The same selection of household chemicals to test from lesson 1 with pipettes, spotting tiles, pre-chopped red cabbage				
4	Strength	4	The same selection of household chemicals to test from lesson 1 with pipettes, spotting tiles, UI solution, UI paper, pH probe (if working), small bottle fizzy water, blue litmus paper				
5	рН	5	1M HCI, 1M NaOH, UI solution				
6	Neutralisation	6	UI solution, 0.5M hydrochloric acid (for demo) and bicarbonate solution, 0.1M hydrochloric acid (for class expt), selection of indigestion tablets				
7	Salts		Mg ribbon and 0.5M HCl, molymods or sweetie balancing For sweetie balancing – 1st reaction - each group needs 2 pots of sweets, one colour per pot and 10 of each sweet. 2nd reaction each group needs 3 pots of sweets, one colour per pot and 5 of each plus cocktail sticks to join them together. Worksheet for sweetie balancing				
8	Assessment	8	Quiz sheet Assessment sheet				
S	Ipportive Reading:	2					
Co	mprehension activity	твс					
As	ssessment:						
Kn	nowledge:	20 question n	nultiple choice quiz				

Extended writing task evaluating the use of an energy resource

Application of Knowledge:

Scheme of Learning: Motion and Pressure

lopic Sequence:									
1	2	3	4	5	6	1	8	<u> </u>	
Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence	
			1.54					- <u>_</u>	

Topic Overview:

In KS2 (primary) pupils are taught to identify the effects of air resistance, water resistance and friction. We build on this with how these forces act between moving surfaces and recognise that some mechanisms including levers, pulleys and gears allow a smaller force to have a greater effect and compare how things move on different surfaces. In year 7 we have covered the contact and non contact forces as well as the affect of resultant forces on motion. This topic is designed to build upon the knowledge of forces and look at how they affect many different areas of science and how we can use forces to calculate pressure in solids, liquids and gasses as well as moments (the turning force sometimes called torque). The whole topic is set to the context of transport and more specifically global parcel delivery services. Students will have to develop their skill to be able to calculate and analyse data in both numeric and graphical forms.

Lesson Sequence:

Supportive Reading:

This topic leans heavily on calculations and analysis of data. Pupils will start by being asked to remember what they learned during the forces topic in year 7 and state the three effects of a resultant forces on an object. We will move on in that lesson to applying the speed equation to calculate speeds, distances and times. This is then developed by looking at how speed can pe represented on a distance/time graph where pupils will need to be able to identify and explain how the gradient of the line represents the speed of the object. We will then look at how forces are also linked to pressure and use the pressure = force/area equation to identify the factor that affects pressure and what happens to objects under high and low pressures. We will develop this further by looking at pressures in fluids and developing our understanding of why some objects will float and other objects will sink. Pupils will then look at pressures in gasses and explain how altitude affects pressure on aircrafts. Pupils will then move on to calculating moments and using the principle of moments to explain why a crane can stay balanced when picking up heavy objects. Finally they will look at how when forces are applied to a spring, work is done and energy is stored in that stretched or compressed object until the force is removed. Pupils will be assessed by 20 multiple choice questions and then required to work out some calculations in a longer answer question.

Se	juence of Lessons:	Resources:				
1	Speed	1				
2	Distance/Time graphs	2	Graph work sheet			
3	Pressure in Solids	3	Ice road truckers print out			
4	Pressure in liquids	4				
5	Pressure in gases	5				
6	Moments (a) Calculation and Principle of moments	6	Class set: Moments rulers and 0.1N Weights and mass hangers			
1	Moments (b) application of knowledge.	ر	Class Set: Meter rulers, string, plastic cups, 1 x pan balances 0.1N Weights and mass hangers.			
8	Work done	8	Class set Springs			
9	Assessment	9	Assessment Sheets			

Comprehension activity	ТВС
Assessment:	
Knowledge:	20 question multiple choice quiz
Application of Knowledge:	Extended Calculation task

Scheme of Learning: Photosynthesis and Respiration

123456789Acids & AlkalisMotion & PressurePhotosynthesis & RespirationMetals & MaterialsWavesInheritance & EvolutionEarth & AtmosphereSpaceEcosystems Interdepender	Topic Sequence:									
Acids & AlkalisMotion & PressurePhotosynthesis & RespirationMetals & MaterialsWavesInheritance & EvolutionEarth & AtmosphereSpaceEcosystems Interdependent	201	2	3	4	5	6		8 o'	9	
	Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence	

Topic Overview:

The national curriculum requirements for this topic are to cover the following:

- The reactants in, and products of, photosynthesis, and a word summary for photosynthesis
- The dependence of almost all life on Earth on the ability of photosynthetic organisms, such as plants and algae, to use sunlight in photosynthesis to build organic molecules that are an essential energy store and to maintain levels of oxygen and carbon dioxide in the atmosphere
- The adaptations of leaves for photosynthesis.
- Aerobic and anaerobic respiration in living organisms, including the breakdown of organic molecules to enable all the other chemical processes necessary for life
- A word summary for aerobic respiration
- The process of anaerobic respiration in humans and micro-organisms, including fermentation, and a word summary for anaerobic respiration
- The differences between aerobic and anaerobic respiration in terms of the reactants, the products formed and the implications for the organism.

Lesson Sequence:

First pupils are introduced to the ideas of how plants absorb light and photosynthesise. This is then linked to how leaves are adapted to allow this process to take place efficiently, involving leaf cell types, organelles and stomata. Pupils then move on to respiration and the different types and situations this takes place in, aerobically, anaerobically and in yeast.

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Sec	juence of Lessons:	Res	esources:		
1	Photosynthesis	1	Ethanol, leaves, kettles, white tiles, iodine.		
2	Structure of leaves	2	Need to print leaf structure worksheet. Leaf structure		
3	Stomata	-	card sort in drawers.		
4	Plants for Food	3	Leaves with nail varnish on underside, microscopes, microscope slides, cover slips. Graph paper, could print pyramids of biomass worksheet		
5	Aerobic Respiration				
6	Anaerobic Respiration in Humans	4			
1	Fermentation in Yeast	5	Limewater, straws, small mirrors		
8	Assessment	6 n/a			
	*	1	Yeast solution, sugar solutions (high, medium and low concentrations), marker pens to write on test tubes, kettles. Could print results table. Graph paper.		
		8	Print MCQ and 6-mark question on yellow paper		

upportive Reading:						
Comprehension activity	твс					
Assessment:						
Knowledge:	20 question multiple choice knowledge test					
Application of Knowledge:	Extended written answer comparing photosynthesis and respiration.					

Scheme of Lea	arning: Metal	ls & Materials		6		2.0	b d		
Topic Sequence:									
¢ 1	2	3	4	5	6	2 7	8	9	
Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence	
Tonio Quorviou				0-0		alle		6	

Topic Overview:

At KS2 students should have begun to classifying materials and link material properties to their uses for some everyday substances. In this topic we cover:

representing chemical reactions using formulae and using equations

combustion, thermal decomposition, oxidation and displacement reactions

reactions of acids with metals to produce a salt plus hydrogen

the order of metals and carbon in the reactivity series

the use of carbon in obtaining metals from metal oxides

properties of ceramics, polymers and composites (qualitative).

Lesson Sequence:

The topic begins with three lessons looking at how different metals behave in three common reactions (with water, acid and oxygen). The results of these experiments are then used to investigate the reactivity series and displacement reactions, including the use of carbon to extract metals from their ores.

The last two lessons look at more complicated materials, covering the properties and uses of ceramics, composites and polymers.

Please note: some of the lessons may take more than the 1 hour lesson slot. Please account for this in your advanced planning.

S	equence of Lessons:	Res	ources:					
1	Metals and acids	1	Mg, Zn, Cu, Al, Fe pieces and 0.5M HCl, white boards					
2	Metals and oxygen	2	Rusty nail, new nails, oil, anhydrous calcium chloride, salt. Mg, Fe, Cu, Zn, Al pieces Balancing equations – version of ppt slide to stick in and balance					
3	Metals and water	3	demo - lithium, sodium, calcium, UI, trough; Class set - Mg, Fe, Cu, Zn, Al,					
4	Reactivity and displacement	I	Demo - copper wire, silver nitrate; Class set - Mg, Zn, Al, Cu and corresponding sulphates, laminated tables for experiment. Reactivity series to stick in; copy of laminated sheet for results					
5	Reactivity and metal extraction	5	Iron oxide and carbon powder, tin lid, magnet					
6	Ceramics and composites	6	Ceramic examples - roof tiles, bricks, pottery, porcelain, bone china Making test bars: cement, sand, aggregate, giant paperclips unbent into rods, card template for former, plasticine					
7	Testing composites and polymers		Demo - making nylon concrete bars – reinforced and unreinforced. Use a mix of 1:3:6 cement:sand:aggregate and unbent giant paperclips for the reinforced bars (or student's prepared bars from last lesson), 100g masses Class set – at least 2 types of plastic bags, hole punch/bulldog clips, Sellotape, 100g masses and hanger					
8	Assessment	8	Quiz sheet Assessment sheet					
S	unnortive Reading.	10						
Supportive nearing.								
Comprehension activity								
٨	reacement.	<u></u>						
K	əəvəəmuyun.	20 question (multiple choice quiz					
	iomouyo.	20 question i						
Application of Knowledge: Extend			iting task evaluating the use of an energy resource					

Scheme of Learning: Waves

Topic Sequence:

1	2	3	4	5	6	1	8	9 💊
Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence

Topic Overview:

Light and sound have been introduced at KS2 with student covering the concepts of sound being produced by vibrations and that light travels in straight lines.

In this topic we cover:

Water waves

Sound waves – properties, production and detection

Light waves – transmission (including the ray model), properties, detection and colour

Lesson Sequence:

The topic starts with a general look at waves including the key vocabulary needed for later in the topic. It moves on to focus on the properties of sound waves, the human ear and applications such as microphones and loudspeakers. The second part of the topic looks at light in more detail, including ray diagrams for reflection and refraction, the human eye, a pinhole camera, lenses and colour.

Please note: some of the lessons will take more than the 1 hour lesson slot. Please account for this in your advanced planning.

Sec	juence of Lessons:	Resources:			
1	Properties of waves	1	Slinky, wave diagram worksheet		
2	Sound waves	2	String telephones, loudspeaker, CRO and sig. gen., pitch and volume worksheet		
3	Hearing	3	Ear model, loudspeaker, CRO and sig.gen., ear diagram worksheet and how we hear worksheet		
4	Controlling sounds	4	Wooden blocks, trundle wheel (or tape measures), microphone worksheet		
5	Using sound	5	Card template, scissors, tape, half toilet roll tubes, 2m lengths of insulated copper wire, homemade speaker demo		
6	Light	6	LDR, desk lamp, multi-meter, leads, selection of materials e.g. paper, card, plastic, tracing paper, wood etc		
1	Ray diagrams – reflection	1	Ray box, slit, power supply, mirror		
8	Ray diagrams – refraction	8	Ray box, slit, power supply, glass block, convex lenses, demo of mug and coin trick		
9	Detecting light	9	Pin hole camera, convex lenses, light source, eye diagram sheet, parts to functions sheet, possible eye dissection		
10	Colour	10	Prisms, filter glasses		
11	Assessment	11	Quiz sheet Assessment sheet		
C	mostive Deading.	2			

ТВС							
20 question multiple choice quiz							
Extended writing task evaluating the use of an energy resource							

Scheme of Learning: Inheritance & Evolution Topic Sequence:									
Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence	
Topic Overview	V:		200		0	8	- 7	1	
The national	curriculum	requirements fo	or this topic a	re to cover t	the following:	no generation	to the next	1	

- neredity as the process by which genetic information is transmitted from one generation to the next
 a simple model of chromosomes, genes and DNA in heredity, including the part played by Watson, Crick, Wilkins and Franklin in the development of the DNA model
- differences between species
- the variation between individuals within a species being continuous or discontinuous, to include measurement and graphical representation of variation
- the variation between species and between individuals of the same species meaning some organisms compete more successfully, which can drive natural selection
- changes in the environment which may leave individuals within a species, and some entire species, less well adapted to compete successfully and reproduce, which in turn may lead to extinction
- the importance of maintaining biodiversity and the use of gene banks to preserve hereditary material

Lesson Sequence:

In the Cells and Organisation and the Reproduction topics in Year 7, pupils learned that DNA is stored in the nucleus of cells and is passed on from parents to offspring at fertilisation. This topic builds upon this knowledge by first looking more closely at the structure and discovery of DNA before introducing the broader idea of evolution through the process of natural selection. We then finish the topic by exploring the causes of extinction. In this way, we move from the microscopic world through to large-scale, macroscopic concepts that affect whole species.

Sec	juence of Lessons:	Resources:			
1 DNA Structure		1	DNA origami template , (make sure you have enough		
2	DNA Discovery	•			
3	Extracting DNA				
4	Genetic Vs Environmental Variation	3	Salt, Washing up liquid, Ethanol – ice cold, Kiwis, Knives, Pestle & mortars, White tiles, Pineapple juice,		
5	Continuous Vs Categoric Variation		Water bath at 37°C.		
6	 6 Species & Classification 7 Natural Selection 8 Extinction 		Chromosome cards		
7			Worksheets in shared area		
8			Shark key and pictures.		
9	Assessment	1	Worksheets in shared area		
1	\geq \circ \circ \circ \circ \circ	8	Worksheets in shared area		
	• • • • • • • •	9 Endangered species factsheets10 Worksheets in shared area			
Sup	portive Reading:	0			
Com	prehension activity Endangered species fact sheets				
Ass	essment:				

ssessment:							
Knowledge:	20 question multiple choice knowledge test						
Application of Knowledge:	Extended written answer explaining elephant evolution.						

Scheme of Learning: Earth & Atmosphere

Topic Sequen	iopic Sequence:							
PU 1	2	3	7 -4	5	6	E 1	8	9
Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence

Topic Overview:

The key stage three national curriculum requires the following content to be taught:

- the composition of the Earth
- the structure of the Earth
- the rock cycle and the formation of igneous, sedimentary and metamorphic rocks
- Earth as a source of limited resources and the efficacy of recycling
- the carbon cycle
- the composition of the atmosphere
- the production of carbon dioxide by human activity and the impact on climate

Lesson Sequence:

We begin by studying the structure and composition of the Earth by describing the crust, mantle and inner and outer core. We include the composition of the atmosphere at this point. We the move onto the rock cycle by describing the properties and formation of sedimentary, igneous and metamorphic rocks and the processes that transform one rock type to another. We then study how carbon is cycled around the environment by natural and artificial processes before looking at the impact humans have on the environment with respect to the importance of recycling and the causes and consequences of climate change.

	0 0 2	0	
S	equence of Lessons:	Res	ources:
1	Structure of the Earth	1	Laminated Earth Structure Fact Sheets
2	Sedimentary Rocks	2	Examples of sedimentary rocks
3	Igneous & Metamorphic Rocks	3	Molten salol, 30x cold slides from freezer, 30x hot slides. Examples of
4	The Rock Cycle	2	ligneous and metamorphic rocks
5	The Carbon Cycle	4	White, dark, milk chocolate, cheese garter, kitchen foil.
6	Recycling	5	n/a
7	Climate Change	6	n/a o-o
8	Assessment	1	n/a
	R HIV	8	Test in shared folder

Supportive Reading:			m		0		
Comprehension activity	твс						
000		0	U	U	0-5-0	1 UN	
Assessment:		100					1
Knowledge:	20 question multip	le choice quiz					
Application of Knowledge:	Extended writing ta	ask					

Scheme of Learning: Space

Topic Sequen	ce:							
1	2	3	4	5	6	1	8	<u> </u>
Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence
		17 0 PA 8 8						

Topic Overview:

In KS2 (primary) pupils are taught to observe changes across the 4 seasons, describe weather associated with the seasons and how day length varies. They have also described the movement of the Earth and other planets relative to the Sun in the solar system and described the movement of the Moon relative to the Earth. This topic in KS3 builds on these principles as well and embedding some of the forces topic that pupils learned in year 7. Space is only covered by Separate science in GCSE so this is potentially the only experience of space some pupils will gain in secondary school. Therefore the depth and detail that is covered is quite extensive even looking at the UKs role in developing satellite communication. We will look at theories about the solar system, how they have changed over time and the evidence that has caused us to adapt our models and developed our understanding.

Lesson Sequence:

This topic leans heavily on literacy and pupils' ability to identify key information within a text and apply it to a context. It starts with an extract by Stephen Hawking on why humans will eventually have to leave the Earth. This is built upon by looking at important aspects for life on Earth, seasonal changes, the phases of the moon and eclipses. Our journey them moves into space looking at natural and artificial satellites. We then look further into space and identify the planets in the solar system, their order and specific features. Astronomical distances are explained at this point to allow pupils to understand the vast distances between planets, stars and galaxies. We eventually look at a satellite that could have life and evaluate the evidence to support the theory that life could be present. Pupils will be required to apply all the knowledge they have learned across many topics during the year to be able to complete the evaluation. We finish the topic with an assessment where the pupils will be asked to explain why the Earth has different seasonal variations in the northern hemisphere, southern hemisphere and at the equator. We include how satellites have allowed us to see the difference in the seasons in the different hemispheres.

1The EarthLamp, Globe, Thermographic film Model of Earth and Sun Worksheet 1: Leaving Earth Worksheet 2: Day length and temperature analysis2Natural Satellites2Lamp and Tennis ball Worksheet 1: Required reading. Worksheet 2: keyword match up print slide 4.3Artificial Satellites3Worksheet 1: British Satellites Worksheet 2: Types of orbits4The history of the solar system3Worksheet 1: British Satellites Worksheet 2: Types of orbits5The Universe and Enceladus4Toilet roll or 3m long strips of paper, Colouring pens Orrery (VI) Worksheet 1: Planet information Worksheet 2: Scale model distances.	Se	quence of Lessons:	Res	SOURCES:
2 Natural Satellites Worksheet 2: Day length and temperature analysis 3 Artificial Satellites Lamp and Tennis ball 3 Artificial Satellites Worksheet 1: Required reading. 4 The history of the solar system Worksheet 1: British Satellites 5 The Universe and Enceladus Toilet roll or 3m long strips of paper, Colouring pens 0// Worksheet 1: Planet information Worksheet 1: Planet information	1	The Earth		Lamp, Globe, Thermographic film Model of Earth and Sun Worksheet 1: Leaving Earth
3 Artificial Satellites 2 Worksheet 1: Required reading. Worksheet 2: keyword match up print slide 4. 4 The history of the solar system 3 Worksheet 1: British Satellites Worksheet 2: Types of orbits 5 The Universe and Enceladus 4 Toilet roll or 3m long strips of paper, Colouring pens Orrery (VI) Worksheet 1: Planet information Worksheet 2: Scale model distances.	2	Natural Satellites -	2	Worksheet 2: Day length and temperature analysis Lamp and Tennis ball Worksheet 1: Required reading
4 The history of the solar system 3 Worksheet 1: British Satellites Worksheet 2: Types of orbits 5 The Universe and Enceladus 4 Toilet roll or 3m long strips of paper, Colouring pens Orrery (VI) Worksheet 1: Planet information Worksheet 2: Scale model distances.	3	Artificial Satellites	2	Worksheet 2: keyword match up print slide 4.
5 The Universe and Enceladus Toilet roll or 3m long strips of paper, Colouring pens 0 Orrery (VI) Worksheet 1: Planet information Worksheet 2: Scale model distances.	4	The history of the solar system	3	Worksheet 1: British Satellites Worksheet 2: Types of orbits
	5	The Universe and Enceladus	4	Toilet roll or 3m long strips of paper, Colouring pens Orrery (VI) Worksheet 1: Planet information Worksheet 2: Scale model distances
6 Assessment 5 Evidence cards and evaluation table for evaluation.	6 A	Assessment	5	Evidence cards and evaluation table for evaluation.
6 Assessment sheet to print.			6	Assessment sheet to print.

Supportive Reading:

Comprehension activity	твс
Assessment:	ANY IS A THE THE AND A THE
Knowledge:	20 question multiple choice quiz
Application of Knowledge:	Extended writing taskExplain why the Earth has different seasonal variation in the northern hemisphere, southern hemisphere and at the equator. You should include how satellites have allowed us to see the difference in the seasons in the different hemispheres.

Scheme of Learning: Ecosystems & Interdependence

Topic Sequence:												
<u> </u>	2	3	• 4	5	6	1	°8° (9				
Acids & Alkalis	Motion & Pressure	Photosynthesis & Respiration	Metals & Materials	Waves	Inheritance & Evolution	Earth & Atmosphere	Space	Ecosystems & Interdependence				
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Topic Overview:

The national curriculum requirements for this topic are to cover the following:

- the interdependence of organisms in an ecosystem, including food webs and insect pollinated crops
- the importance of plant reproduction through insect pollination in human food security.
- how organisms affect, and are affected by, their environment, including the accumulation of toxic materials.

Lesson Sequence:

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Starting with how animal and plants are adapted to their environments to help them compete, we move onto food webs and predator prey cycles. Then look at how organisms affect, and are affected by, their environment, including the accumulation of toxic materials and our dependence on them for our food security. We include a lesson on sampling techniques as a foundation for later KS4 work.

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Sei	juence of Lessons:	Resources:				
1	Competition in Animals	1.	n/a			
2	Competition & Adaptation in Plants	2	n/a			
3	Adaptation in Animals	3	Polar adaptation comprehension activity			
4	Food Webs	4	Ocean food webs cut & stick			
5	Predator Prey Cycles	5	Predator Prey Graphs			
6	Bioaccumulation	6	Sorting sheet and video questions			
7	Sampling Techniques	7	Quadrats and trundle wheel			
8	Food Security	8	Food security comprehension activity			
9	Assessment Lesson	9	In folder			
111		100				

Supportive Reading:	
Comprehension activity	Polar adaptations L3 and Food Security L8
V. int	
Assessment:	
Knowledge:	20 question multiple choice knowledge test
Application of Knowledge:	Extended written answer describing sampling techniques.

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Scheme of Learning: Cell Biology

Topic Sequence:

12345678910Cell BiologyParticle Model of MatterInfection & ResponseAtomic Structure & the Periodic TableAtomic Structure (Physics)Bonding & StructureEnergyBioenergeticsRates of ReactionChemistry of the Atmosphere										
Cell BiologyParticle Model of MatterInfection & ResponseAtomic Structure & the Periodic TableAtomic Structure (Physics)Bonding & StructureEnergyBioenergeticsRates of ReactionChemistry of the Atmosphere	്ര1 🦯	2	3	4	5	6	1	8	9	10
	Cell Biology	Particle Model of Matter	Infection & Response	Atomic Structure & the Periodic Table	Atomic Structure (Physics)	Bonding & Structure	Energy 6	Bioenergetics	Rates of Reaction	Chemistry of the Atmosphere

Topic Overview:

Cells are the basic unit of all forms of life. In this section we explore how structural differences between types of cells enables them to perform specific functions within the organism. These differences in cells are controlled by genes in the nucleus. For an organism to grow, cells must divide by mitosis producing two new identical cells. If cells are isolated at an early stage of growth before they have become too specialised, they can retain their ability to grow into a range of different types of cells. This phenomenon has led to the development of stem cell technology. This is a new branch of medicine that allows doctors to repair damaged organs by growing new tissue from stem cells.

Lesson Sequence:

We start by describing the structure and function of generic plant and animal cells and then specialised cells. We then look at the differences between prokaryotic and eukaryotic cells before moving on to the genetic information cells have. There are two required practicals in this topic (magnification and aseptic Technique). We then describe the process of mitosis and finish with a comparison of the different methods of cell transport. Our focus skill for this topic is 'planning a method'.

Seq	uence of Lessons:	Res	jources:					
1	Animal and Plant Cell Differences		Microscope diagram, Cells labelling worksheet – both found in shared area. Microscopes, pre-prepared slides of basic animal and plant cells.					
2	Microscopes <i>Required Practical</i> n topic assessment (likely 2 lessons)	nid 2	Microscopes, onion (cut into pieces), tweezers, iodine, slides, cover slips. Required Practical method sheet to print – in shared area. Mid topic assessment in shared area. Practice questions in shared area.					
3	Prokaryotes and Eukaryotes. Litera opportunity	cy 3	Diagrams of example eukaryote and prokaryote printed and in drawers. Could print table of differences for pupils to complete. Collect literacy sheet to read from drawers.					
4	Specialised Cells	4	Table for pupils to complete on PowerPoint. Information sheets in drawers. Post it notes for					
5	Aseptic Technique Required Practical.		plenary. Print exam question if needed.					
6	Genetic Information		Print RP sheets, print example plates for pupils to measure. Nutrient agar plates spread with bacteria, 3 types of filter paper discs soaked in mouthwash or TCP or antiseptic, disinfectant					
7	Mitosis	2-8	for desks, tape, forceps, pens. DEMO – agar plate, 'bacteria' culture, glass spreader, inoculating loop.					
8	Stem cells (2 lessons)	6	Print diagram and handout for pupils of chromosomes and paragraph. Can print and use					
9	Evaluation of therapeutic cloning	<u> </u>	DNA model origami if desired.					
10	Diffusion- Concentration and		Could print table for pupils to fill out details of cell cycle. Print exam question.					
10	temperature	8	Resources in shared area					
11	Diffusion – Surface area	9	People cards for ethics task to be printed and given to pupils in groups. Could use IT room.					
	How does the concentration of sug	ar 10	Blocks of agar coloured with phenylalanine, 0.5M HCl, 1.0M HCl, water bath at 50 degrees. Print results table slide.					
12	Practical - mid topic assessment -	11	Demo – icing sugar, sugar cube in Bunsen flame. Print table for pupils to complete. Info sheets in drawer.					
13	Active Transport		0.2, 0.4, 0.6, 0.8, 1.0M sugar solutions, cylinders of potato, high resolution balances. Print results table if needed.					
14	Revision	13	Print root hair cell diagram to annotate and exam questions.					
15			Question mat to print					
		- 15	Test paper in shared area					
Sup	portive Reading:	/						
Com	prehension activity	From Pro	karyotes to Eukaryotes information to read					
Asse	essment:							
Know	wledge:	Multiple	choice and short answer questions.					
Anni	ication of Knowledge	Exam que	estions based on the skill of 'plan a method'					

Scheme of Learning: Particle Model of Matter

I ohic 2ednei	ICE:								
1	2	<u> </u>	4	5	6	1	8	9	10
Cell Biology	Particle Model of Matter	Infection & Response	Atomic Structure & the Periodic Table	Atomic Structure (Physics)	Bonding & Structure	Energy	Bioenergetics	Rates of Reaction	Chemistry of the Atmosphere

Topic Overview:

The particle model is widely used to predict the behaviour of solids, liquids and gases and this has many applications in everyday life. It helps us to explain a wide range of observations and engineers use these principles when designing vessels to withstand high pressures and temperatures, such as submarines and spacecraft. It also explains why it is difficult to make a good cup of tea high up a mountain

Lesson Sequence:

We begin with a recap of the states of matter and changes of state from key stage three. We then build on this foundation to look deeper at what internal energy of a substance is and how potential and kinetic energy of particles changes during temperature rises and state changes. We include the required practical on specific heat capacity.

The middle part of the topic concerns density. We complete another required practical to calculate the densities of regular shapes, irregular shapes and liquids. We finish by looking at gas pressure.

In this topic, we will focus on the skill of 'describe'.

Sec	juence of Lessons:	Res	Resources:				
1	States of Matter	1	n/a				
2	Change of State	2	Melted stearic acid in test tubes in a water bath				
3	Internal Energy	3	n/a				
4	Specific Heat Capacity (2 lessons worth) (required practical)	4	Demo: Specific heat capacity required practical – use a joule meter, not an ammeter and voltmeter.				
5	Specific Latent Heat	5	n/a				
6	Density of Regular Shapes (required practical)	6	Various metal cubes, callipers.				
7	Density of Irregular Shapes (required practical)	1	Variety of irregular shaped objects, eureka cans,				
1/	Density of Liquids (required practical) Mid topic		washing up bowls.				
8	assessment	8	Variety of liquids (golden syrup, wallpaper paste, sugar				
9	Gas Pressure & Temperature		water, salt water, glycerol, sunflower oil) balances.				
10	Revision	9	Empty drink cans, washing up bowls.				
11	Test	10	n/a				
		11	n/a				

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Supportive Reading:	$(\Psi) = \{\chi_{ijj} \in \Psi \} \in [X_i]$		
Extended answer practice	8 marker written answer in lesson 3		
/ # 1 / m A			
Assessment:		- M - 21 - 20 - 20 - 20 - 20 - 20 - 20 - 20	
Knowledge:	Multiple choice and short answer questions.	1	
Application of Knowledge:	Exam questions based on the skill of describe	$\sim \rightarrow \sim$	

Scheme of Learning: Infection & Response

Topic Sequence: 1 2 3 4 5 6 7 8 9 10 Particle Atomic **Chemistry of** Atomic **Infection & Bonding & Rates of Cell Biology Model of** Structure Energy **Bioenergetics** Structure & the the Structure Reaction Response **Periodic Table** Matter (Physics) **Atmosphere**

Topic Overview:

Pathogens are microorganisms such as viruses and bacteria that cause infectious diseases in animals and plants. They depend on their host to provide the conditions and nutrients that they need to grow and reproduce. They frequently produce toxins that damage tissues and make us feel ill. This section will explore how we can avoid diseases by reducing contact with them, as well as how the body uses barriers against pathogens. Once inside the body our immune system is triggered which is usually strong enough to destroy the pathogen and prevent disease. When at risk from unusual or dangerous diseases our body's natural system can be enhanced by the use of vaccination. Since the 1940s a range of antibiotics have been developed which have proved successful against a number of lethal diseases caused by bacteria. Unfortunately many groups of bacteria have now become resistant to these antibiotics. The race is now on to develop a new set of antibiotics

Lesson Sequence:

Starting with an overview of communicable disease and their transmission, including a description of microbes, we learn about the specific disease required by the specification. We then describe the immune response of the human body and how this can be supported with antibiotics and vaccinations. We explore how new medicines are developed and the use of monoclonal antibodies. We finish with a look at how plants protect themselves from pathogens. Our focus skill for this topic is 'describe'.

		- Becources-				
Seq	UENCE OT LESSONS:					
1	Communicable disease	1	Virus, fungi, bacteria & protist fact sheets – laminated in			
2	Spread of Disease.	-	ning cabinet			
3	Viral & bacterial disease	2	Petri dishes with sterile agar, cotton buds, tape, marker			
4	Fungi & protist disease Defence mechanisms of human body.		Viral and hacterial disease laminated fact sheets in filing			
5			cabinet			
6	Vaccination. Literacy task	4	Worksheets in shared area folder			
7	Antibiotics & painkillers	5	Worksheets in shared area folder			
8	Developing medicines	6	Vaccination and covid literacy task sheets.			
9	Double blind trials & placebos - Mid topic assessment	7	Patient card and action cards in filing cabinet.			
10	Monoclonal antibodies	8	Worksheets in shared area folder			
11	Plant diseases		Two number grids x 15. Two colours of pre-wrapped (for			
12	Revision	्	hygiene) sweets. Exam question in shared area folder.			
13	Test	10	Monoclonal antibody laminated fact sheets			
1		11	Symptom checker question key. Laminated pictures of diseased plants.			
		12	Resources in shared area folder			
1		13	Test paper in shared area			
1000						

upportive Reading:								
Comprehension activity Vaccinations & Covid comprehension L6								
Assessment:								
Knowledge:	Multiple choice and short answer questions.							
Application of Knowledge:	Exam questions based on the skill of 'describe'							

Scheme of Learning: Atomic Structure and the Periodic Table

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1	2	3	7-4	5	6	1	8	9	10
Cell Biology	Particle Model of Matter	Infection & Response	Atomic Structure & the Periodic Table	Atomic Structure (Physics)	Bonding & Structure	Energy	Bioenergetics	Rates of Reaction	Chemistry of the Atmosphere
Tonic Overvie	w:				161	51	a Y		

The periodic table provides chemists with a structured organisation of the known chemical elements from which they can make sense of their physical and chemical properties. The historical development of the periodic table and models of atomic structure provide good examples of how scientific ideas and explanations develop over time as new evidence emerges. The arrangement of elements in the modern periodic table can be explained in terms of atomic structure which provides evidence for the model of a nuclear atom with electrons in energy levels.

Lesson Sequence:

We begin by building upon key stage three knowledge of elements, compounds and mixtures and experience a variety of separation techniques including filtration, chromatography, distillation and crystallisation.

We then look more closely at the atom, learning about sub-atomic particles and the scientific discoveries that lead to our current understanding of the atom's structure. We introduce the concept of atomic mass and isotopes before discovering how the periodic table came into existence through the work of Mendeleev and other scientists.

Once we have introduced ions and how the metals and non-metals vary in the ions they form, we spend the next few lessons focusing on groups in the periodic table. We look at the properties and trends of groups 1, 7, 0 and the transition metals.

The focus skill for this topic is 'compare'. We do this by comparing the groups in the periodic table, and the different models of the atom through history.

Sea	uence of Lessons		Resources:					
1	Elements & (Compounds	1	Worksheets in shared area folder				
2	Chemical Eq	uations & Mixtures	5	3 Demos: top pan balance, NaOH + $CuSO_4$, vinegar & baking powder, zip lock bag, approx. 4g iron wool, 4g plasticine, metro stick and fulcrum tin tray (e.g. takeaway carton)				
3	Filtration & 0	Crystallisation	2					
4	Distillation &	Chromatography	2	Class practical: rack salt				
5	History of th	e Atom	-					
6	Atomic Struc	ture	4	Demo: Simple distillation of ink and water. Class practical: Chromatography – variety of pens, paperclips, chromatography paper slips.				
1	Relative Ator	nic Mass & Isotopes	19					
8	Developmen	t of the Periodic Table mid-topic assessment	5	Worksheets in shared area folder				
9	lons, Metals	& Non-Metals	6	Worksheets in shared area folder				
10	Group 1 mid	-topic assessment	1	Worksheets in shared area folder				
11	Group 7		8	n/a				
12	Group 0 & Ti	ansition Metals	9	n/a				
13 14	Revision Test		10	Group 1 metals, water trough & lid, universal indicator, scalpel, white tile				
Sup	Supportive Reading:			Dimple tiles, pipettes, Chlorine, bromine & iodine water, potassium chloride, bromine & iodide solutions				
Lite	writing longer answer opportunities in various lessons and peer marking		5 12	sodium hydroxide solution, solutions of Cu ²⁺ , Fe ²⁺ , Fe ³⁺ and transition metal sheets in filing cabinet				
Ass	issessment:		13	Resources in shared area folder				
Kno	wledge:	Multiple choice and short answer questions.	14	Test in shared area folder				
App Kno	lication of wledge:	Exam questions based on the skill of 'compare'.		300 g again				

Scheme of Learning: Atomic Structure (Physics)

iohic gedinence.									
1	2	3	4	5	6	1	8	9	10
Cell Biology	Particle Model of Matter	Infection & Response	Atomic Structure & the Periodic Table	Atomic Structure (Physics)	Bonding & Structure	Energy	Bioenergetics	Rates of Reaction	Chemistry of the Atmosphere

Topic Overview:

Ionising radiation is hazardous but can be very useful. Although radioactivity was discovered over a century ago, it took many nuclear physicists several decades to understand the structure of atoms, nuclear forces and stability. Early researchers suffered from their exposure to ionising radiation. Rules for radiological protection were first introduced in the 1930s and subsequently improved. Today radioactive materials are widely used in medicine, industry, agriculture and electrical power generation.

Lesson Sequence:

We begin with a recap of the history of the atomic structure model and the diagram we use today. This was covered in topic on Atomic Structure and the Periodic Table as it is a fundamental idea behind most of the Chemistry topics and but also many of the Physics topics too. However, at this point we move away from the focus on electron shells and bonding and towards the nucleus with its protons and neutrons.

The topic then moves onto the idea that large, heavy atoms are unstable and will randomly decay to become a different element. This decay causes particles and energy to be emitted either as an alpha particle, a beta particle or a gamma wave (energy), possibly more than one of these at a time. The lessons then go on to look at the effect the emission of these particle has on the original nucleus.

Next in the topic we look at the different ways to measure radioactivity, including background radiation and consider the statistical constant that is the half-life of a radioactive element.

The uses of radioactivity and its associated risks are then considered; from tracers and medical applications to radio-carbon dating. Finally, the processes of nuclear fission and fusion are looked at in more detail.

Se	quence of Lessons:	Resources:				
1	Structure of the Atom	1	n/a			
2	Radioactive Decay	2	Summary Qs and comparison table			
3	Nuclear Equations	3	PT with mass numbers, decay worksheet			
4	Activity & Half-Life	4	Skittles, big measuring cylinders x4, half-life questions			
5	Dangers of Radiation – <i>mid-topic assessed question</i>	5	Background pie chart, exam Q, Geiger-Müller tube,			
6	Uses of Radiation	×	counter			
1	Nuclear Fission & Fusion	6	Uses of Radiation Qs			
8	Revision	7	Fission & Fusion Qs (use the level 3 Qs for top set)			
9	Test	8	Resources in shared folder			
		9	Test in shared folder			

Supportive Reading:		
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Assessment:		
Knowledge:	Multiple choice and short answer questions.	
Application of Knowledge:	Exam questions based on the skill of 'explain'.	AL 1

Scheme of Learning: Bonding & Structure

T	onic	Sen	uence:	
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1	2	3	774	5	6	1	8	9	10
Cell Biology	Particle Model of Matter	Infection & Response	Atomic Structure & the Periodic Table	Atomic Structure (Physics)	Bonding & Structure	Energy	Bioenergetics	Rates of Reaction	Chemistry of the Atmosphere
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Topic Overview:

Chemists use theories of structure and bonding to explain the physical and chemical properties of materials. Analysis of structures shows that atoms can be arranged in a variety of ways, some of which are molecular while others are giant structures. Theories of bonding explain how atoms are held together in these structures. Scientists use this knowledge of structure and bonding to engineer new materials with desirable properties. The properties of these materials may offer new applications in a range of different technologies.

Lesson Sequence:

Application of Knowledge:

We begin with a recap of previous chemistry knowledge by looking at the structure of the atom. It is essential for this topic, and indeed all of Chemistry, that pupils have a firm understanding of what electrons are. We look at how an atom can lose or gain an electron to form an ion. This leads onto how these electrons are transferred in ionic compounds. Pupils need to be able to explain the properties of small and giant ionic compounds.

We then move onto bonding between non-metals where electrons are not transferred, but shared. Again, pupils need to be able to describe the structure and properties of simple and giant covalent structures including diamond, graphite, fullerenes and graphene.

The final type of bonding we study is metallic bonding, Pupils needs to be able to explain why metals conduct heat and electricity with reference to their electrons.

Pupils need to be able to explain how the forces of attraction in solids, liquids and gases affect their properties. Finally, Separate Chemists learn about nanoparticles and their applications.

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Seq	uence of Lessons:	IT to	Resources:						
1	Atoms into lons	ELE I	1	n/a					
2	Ionic Bonding	47	2	n/a					
3	Giant Ionic Structures – <i>mid topic assessment</i>			Salt (NaCl), Carbon electrodes, Crocodile clips, Power					
4	Covalent Bonding		packs, Wires, Bulbs						
5	Structure of Simple Molecules	10	4	n/a					
6	Giant Covalent Structures		5	n/a					
7	Fullerenes & Graphene		6	n/a					
8	Bonding in Metals			n/a					
9	States of Matter			n/a					
10	Nanoparticles – Separate Chemistry Only			n/a					
11	Revision	TA	10	Nanoparticle Info Sheets					
12	Test		11	l n/a					
	Seco (12	Test in shared area					
Sup	portive Reading:	jó.							
Lite	acy tasks	Writing longer answer opport	tuni	ties in various lessons and peer marking					
Ass	essment:								
Kno	wledge:	Multiple choice and short ans	wer	questions.					

Exam questions based on the skill of 'explain'.

Scheme of Learning: Energy

1 2 3 Particle Infections	4					lopic Sequence:									
Particle Infection of		5	6	1	8	9	10								
Cell Biology Model of Response Matter	Atomic Structure & the Periodic Table	Atomic Structure (Physics)	Bonding & Structure	Energy	Bioenergetics	Rates of Reaction	Chemistry of the Atmosphere								

Topic Overview:

The concept of energy emerged in the 19th century. The idea was used to explain the work output of steam engines and then generalised to understand other heat engines. It also became a key tool for understanding chemical reactions and biological systems. Limits to the use of fossil fuels and global warming are critical problems for this century. Physicists and engineers are working hard to identify ways to reduce our energy usage.

Lesson Sequence:

The topic begins with a look at the idea of energy as a means to calculate if it possible for something to happen. This involves looking at the stores of energy and the 'pathways' or energy transfers that occur.

The next part of the topic looks in more detail at how we calculate the amount of energy in the kinetic, gravitational, elastic and thermal energy stores, including the required practical on specific heat capacity.

Next it covers work done and relates this to the transfer of energy/pathways linking with the concepts of power and efficiency. The required practical on insulation fits into this section of the topic by looking at how unwanted energy transfers can be reduced. Finally, the topic moves onto the energy resources used to generate electricity. Finishing with a lesson looking at how exam questions might be related to current environmental issues relating to climate change.

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Seq	uence of Lessons:	Reso	Irces:				
1	Energy stores and pathways	1	Energy analogy demo (coloured water, 2 containers, 2 tubes), worksheet				
2	Kinetic energy	2	Kinetic energy calculation sheet				
3	Gravitational energy	3	Gravitational potential energy sheet				
4	Elastic energy mid-topic assessment	4	Spring toys, Hooke's Law practical – springs, 100g masses and hanger, metre rulers, elastic energy sheet, mid-topic assessment sheet,				
5	Specific Heat Capacity	5	Resource sin shared area				
6	Power	6	1kg of 100g masses, metre ruler, timer, power worksheet				
1	Efficiency mid-topic assessment	7	Sankey diagram sheet, efficiency sheet, Mid-topic assessment sheet				
8	Conduction	8	Conduction box demo and kettle (different rods into metal trough)				
9	Convection (optional)	9	Tea bag demo, potassium permanganate 'tea bag' demo, chimney demo				
10	Insulation (required practical)	10	Class set: insulation practical (kettles, thermometers, beaker lids, different insulating materials, paper towels, elastic bands) +instruction sheet				
11	Energy resources	11	Laminated energy resources sheets, blank summary sheet, poster making resources				
12	Evaluating energy resources	12	'we do' and 'you do' GCSE questions				
13	Revision	13	n/a				
14	Test	14	test				
		11.2					

Supportive Reading:		
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Assessment:		
Knowledge:	Multiple choice and short answer questions.	
Application of Knowledge:	Exam questions based on the skill of calculate	

Scheme of Learning: Bioenergetics

iopic Sequence:											
01 /	2	3	4	5	6	1	8	9	10		
Cell Biology	Particle Model of Matter	Infection & Response	Atomic Structure & the Periodic Table	Atomic Structure (Physics)	Bonding & Structure	Energy 6	Bioenergetics	Rates of Reaction	Chemistry of the Atmosphere		
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Topic Overview:

In this section we will explore how plants harness the Sun's energy in photosynthesis in order to make food. This process liberates oxygen which has built up over millions of years in the Earth's atmosphere. Both animals and plants use this oxygen to oxidise food in a process called aerobic respiration which transfers the energy that the organism needs to perform its functions. Conversely, anaerobic respiration does not require oxygen to transfer energy. During vigorous exercise the human body is unable to supply the cells with sufficient oxygen and it switches to anaerobic respiration. This process will supply energy but also causes the build-up of lactic acid in muscles which causes fatigue.

Lesson Sequence:

We begin with a recap of the Cells topic from the start of year 9 and then focus on chloroplasts and the process of photosynthesis. The required practical for this topic investigates how the rate of photosynthesis is affected by the intensity of light. After the practical, we look at other limiting factors (other than light) and how they can be controlled in greenhouses to increase the rate of photosynthesis in plants grown there. We conclude the plants lessons by describing how plants use the glucose they produce during photosynthesis.

The second half of the topic looks at respiration. Beginning with aerobic respiration, we then explain the changes that occur in the body during exercise and compare aerobic to anaerobic respiration. We conclude the topic by learning about metabolism and the role of the liver.

equence of Lessons:	Resources:						
Photosynthesis	1 Exam question in shared folder						
The Rate of Photosynthesis – Required Practical	2 Pondweed, lamps, metre sticks, glass funnels, large beakers,						
Limiting Factors	sodium hydrogen carbonate solution, plasticine.						
The Uses of Glucose	3 Worksheets in shared area						
	4 Worksheets in shared area						
Aerobic Respiration – <i>mid topic assessment</i>	5 Limewater, straws						
The Response to Exercise	6 n/a						
Anaerobic Respiration	7 n/a						
Metabolism and the Liver	8 n/a						
Revision	9 Resources in shared area						
Test	10 Test in shared area						

Supportive Reading:		1.0 0
Literacy tasks	Response to exercise lesson has a longer written answer to complete	
		$\langle \sigma \cap \gamma \rangle$
Assessment:		
Knowledge:	Multiple choice and short answer questions.	
Application of Knowledge:	Exam questions based on the skill of 'explain'	

Scheme of Learning: Rates of Reaction

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۶///1	2	3	14	5	6	1.8	8	9	10
Cell Biology	y Particle Model of Resp Matter	Particle Infection & Atomic Atomic Model of Infection & Structure & the Structure Matter Response Periodic Table (Physics)	Bonding & Structure	Energy	Bioenergetics	Rates of Reaction Atm	Chemistry of the Atmosphere		
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Topic Overview:

Chemical reactions can occur at vastly different rates. Whilst the reactivity of chemicals is a significant factor in how fast chemical reactions proceed, there are many variables that can be manipulated in order to speed them up or slow them down. Chemical reactions may also be reversible and therefore the effect of different variables needs to be established in order to identify how to maximise the yield of desired product. Understanding energy changes that accompany chemical reactions is important for this process. In industry, chemists and chemical engineers determine the effect of different variables on reaction rate and yield of product. Whilst there may be compromises to be made, they carry out optimisation processes to ensure that enough product is produced within a sufficient time, and in an energy-efficient way.

Lesson Sequence:

We begin this topic by defining rate of reaction and how it can be quantified. We then carry out a series of practical investigations to demonstrate how surface area, temperature, concentration and catalysts effect the rate of chemical reactions. We spend a lesson on the maths skill of drawing tangents on a curved line graph to determine rate and then finish the topic focussing on reversible reactions and how changing conditions affects their equilibrium. The focus skill for this topic is graph skills.

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Seq	uence of Lessons:	R	Resources:						
1	Rate of Reaction		1.0mol/dm ³ hydrochloric acid, Calcium carbonate,						
2	Collision Theory & Surface Area	a 13 -	Balances, Cotton wool						
3	Temperature & Rate of Reaction	on	Gas syringes, 1M Hydrochloric acid, Balances, Marble chips, Powdered CaCO ₂						
4	Concentration & Rate – <i>Requir</i> assessment	ed Practical & mid topic	Conical flasks (correct size for gas syringe bungs)						
5	Required Practical Write Up		cream tubs, Kettles						
6	The Effect of Catalysts		Sodium thiosulphate, X's on card, 0.25M, 0.5M, 1M, 2M						
7	Using Tangents		hydrochloric acid						
8	Reversible Reactions		Uncertainty worksheet						
9	Altering Conditions		Hydrogen peroxide, Cooked liver, Raw liver Manganese oxide, Copper oxide, Iron oxide						
10	Revision		Measuring rates worksheet						
	Test	16	Class set: Anhydrous copper sulphate Demo: Traffic light reversible reaction bottle						
_			n/a o_ o_						
Sup	portive Keading:	1	0 Worksheets in shared folder						
TBC	;	1	1 Test in shared folder						
	0	EC 1 Ka	0						
Asse	essment:								
Know	wiedge:	Multiple choice and short answ	ver questions.						

Application of Knowledge:

Exam questions based on the skill of 'graph skills'.

Scheme of Learning: Chemistry of the Atmosphere

Topic Sequence:										
2 1	2	3	4	5	6	1	8	9	10	
Cell Biology	Particle Model of Matter	Infection & Response	Atomic Structure & the Periodic Table	Atomic Structure (Physics)	Bonding & Structure	Energy	Bioenergetics	Rates of Reaction	Chemistry of the Atmosphere	
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The Earth's atmosphere is dynamic and forever changing. The causes of these changes are sometimes man-made and sometimes part of many natural cycles. Scientists use very complex software to predict weather and climate change as there are many variables that can influence this. The problems caused by increased levels of air pollutants require scientists and engineers to develop solutions that help to reduce the impact of human activity.

Lesson Sequence:

In this short topic, we start by exploring the historical changes that have occurred in Earth's atmosphere – from the planet's origin up to the present era. We then look at the effect human activity has had on the atmosphere in terms of climate change and pollution.

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Seq	uence of Lessons:	Res	ources:
1	History of our Atmosphere	1	Card sort
2	Our Evolving Atmosphere	2	Reading comprehension
3	Greenhouse Gases – mid topic assessment	3	n/a
4	Global Climate Change	4	Climate change fact sheets x 9, Answer table.
5	Atmospheric Pollutants	5	n/a
6	Revision	6	n/a
7	Test	7	Test in shared folder

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Supportive Reading:						
Literacy tasks	Comprehension activity in L2	2 and longer w	ritten ans	wer will be assesse	d in L3	
0		2		0		1/2)
Assessment:		0	0		0	
Knowledge:	Multiple choice and short ans	swer question	s.			
Application of Knowledge:	Exam questions based on the	e skill of 'expla	in'.			